PRELIMINARY AMENDMENT Continuation Application of U.S. Appln. No. 08/416,738

Replace the first paragraph of page 13 with the following paragraph:

 $O_{\mathcal{A}}$

A suitable calcination temperature is not necessarily critical since it depends on the kind of the intended metal oxide, the kinds and concentrations of the hydrogen halide, the molecular halogen and the component prepared from the molecular halogen and steam, or the calcination time. It is preferably from 500 to 1500°C, more preferably from 600 to 1400°C. When the calcination temperature is lower than 500°C, a long time is necessary for calcination. When the calcination temperature exceeds 1500°C, many agglomerated particles tend to be contained in the produced metal oxide powder.

Replace TABLE 2 on Page 34 with the following:



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Table 2

Ex.	Oxide							Calci	nation	Calcination conditions		
No.			A	tmos	Atmosphere gas (vol. %)	gas (vol. %			Gas intro-	Maintaining	Maintaining
		HCl	HBr	r $ HF $ Cl_2	Cl_2	N_2	N_2 H_2O H_2 Air	$ m H_2$		duction	temp. (°C)	time (min.)
										temp. (°C)		
11	$Ti0_2$	100								Room temp.	800	30
12	$Ti0_2$	45					10		45	Room temp.	1100	30
13	${ m Ti0_2}$	100								Room temp.	1100	30
14	$\mathrm{Ti0}_2$	100								800	1100	30
15	$\mathrm{Ti0}_{2}$	30				1 0 2				800	1100	30
16	$Ti0_2$	30							02	800	800	30
17	$ m Ti0_2$				30	09	10			800	1100	30
18	$Ti0_2$				100					800	1100	30
19	$ { m Ti0_2} $				30	09	10			800	1100	30
C. 1	$Ti0_2$								100	Room temp.	1100	180
C. 2	$\mathrm{Ti0}_{2}$								100	Room temp.	1100	180